1. MATERIAL NAME

REMELT INGOT AND CAST ALUMINUM ALLOY 369.1

Chemical description

Solid Aluminum Metal Ingot

Manufacturer

SPECTRO ALLOYS INC 13220 DOYLE PATH E ROSEMONT, MN 55068

Telephone

1-651-480-6109

Fax

1-651-438-3714

Emergency Information:

1-651-480-6109

2. HAZARD IDENTIFICATION

Classification

This product is considered hazardardous per 29 CFR 1910.1200

Emergency overview

Solid, silver colored. Odorless. Not combustible as shipped. Small chips, fine turnings and dust from processing are easily Ignitable Explosion/fire hazards will be present

- Dust or fines dispersed in air
- Chips, dust or fines in contact with water
- Dust and fines in contact with certain oxides (Iron oxide, Copper oxide)
- Molten metal in contact with water or some oxides (iron & copper)

Dust and fume from processing can cause respiratory tract irritation

Potential health hazards

Sensitization, respiratory

Sensitization, skin

Carcinogenicity

Category 1

Category 2

Carcinogenicity Category 2
Reproductive Toxicity Category 1A

Specific target organ toxicity, repeated Category 1 (Respiratory tract)

Precautionary statement

Prevention

Obtain special instructions before use. DO NOT BREATH dust/fume In case of inadequate Ventilation wear respiratory protection. Wear protective gloves/clothing/eye protection DO NOT eat, drink or smoke when using this product. Wash thoroughly after handling .this product. Prevent dust accumulation to minimize explosion hazard

Response

If inhaled: If breathing is difficult, remove person to fresh air. If respiratory symptoms are experienced call a doctor at once. IF ON SKIN wash with soap and water. If rash occurs see a doctor promptly. Wash contaminated clothing before reuse

Storage

Store in a dry place

Disposal

Dispose of contents/container in accordance with local State or Federal Regulations

2. HAZARD IDENTIFICATION

Carcinogenicity &

Reproductive Hazard As shipped product no hazard Dust from processing can cause

problems as Cancer (Cadmium Nickel Cobalt Lead) Reproductive hazard (Lead, Manganese) Dust & fumes from welding or elevated temperature

processing can present a cancer hazard (hexavalent chromium compounds, lead compounds, cadmium compounds, cobalt compounds, nickel compounds, welding fumes) can present a reproductive hazard (lead compounds, manganese compounds)

Medical conditions aggravated

Asthma, chronic lung disease,

Secondary Parkinson's disease and skin rash

3. COMPOSITION INFORMATION ON INGREDIENTS

Components	CAS#	Percent
Silicon	7440-21-3	11.0 – 12.0
Aluminum	7429-90-5	>75
Copper	7440-50-8	0.50
Nickel	7440-02-0	0.05
Zinc	7440-66-6	0.90
Manganese	7439-96-5	0.35
Magnesium	7439-95-4	0.30 - 0.45
Chromium	7440-47-3	0.30 - 0.40
Titanium	7440.32-6	
Tin	7440-31-5	0.10
Iron	7439-89-6	1.0
Others Total		0.15

4. FIRST AID MEASURES

Eye contact

Dust & fumes from processing: Rinse eyes with water or saline

for at least 15 minutes. Consult doctor

Skin contact

Dust & fumes from processing Wash with soap & water for at least 15

minutes. Get medical attention if irritation persists

Inhalation

Dust & fumes from processing remove to fresh air. Check for clear airway, breathing and pulse. If breathing is difficult provide oxygen. Loosen tight clothing. Provide CPR for persons without pulse or respiration Consult a

doctor

Ingestion

Not relevant due to nature of product

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4. FIRST AID MEASURES (Continued):

Most important Symptoms/effects

Health effect from mechanical, processing cutting, grinding) Dust from processing can cause chronic over exposure can cause anemia, skin pigment changes, pulmonary

fibrosis.

secondary Parkinson's disease, lung cancer . Can cause anemia, skin pigmentation Acute and delayed changes, pulmonary damage, reproductive harm and central nervous system damage

If contains Cobalt or Nickel may produce allergic reaction

aggravated by exposure

Medical conditions Added health effects aggravated by elevated temperature processing (welding, melting) Dust and fumes from processing Over exposure can cause metal fume fever, pulmonary fibrosis, pulmonary edema, secondary Parkinson's disease, lung

Immediate medical attention

Indication of need If breathing is difficult, administer oxygen

General information If exposed see a doctor

5. GENERAL FIRE HAZARDS

This product as shipped does not present a hazard. Small chips, fine turnings, and dust from processing may be readily ignited.

Extinguishing media Suitable extinguishing media

> Use Class D Extinguishing agents on fines dust or molten metal Use course dry sand to surround molten metal DO NOT USE water on spray chip & turnings

Unsuitable extinguishing media

DO NOT USE halogenated agents on small chips or fines DO NOT USE water in fighting fires around MOLTEN METAL These fire extinguishing agents react with the burning material

Specific hazards arising from

the Chemical

The product does not present fire or explosion hazard as shipped Small chips, turnings and dust from processing is readily ignitable be a potential hazard under following conditions;

- Dust clouds may be explosive. Even a minor dust cloud can explode violently. Dust accumulation on floor, ledges and beams can present a risk of ignition, flame propagation and secondary explosions
- Chips, fines and dust in contact with water can generate flammable hydrogen gases. These gases present an explosion hazard in poorly ventilated spaces

5. FIRE FIGHTING MEASURES (continued)

Dust and fines in contact with certain metal oxides (Iron oxide, copper oxide) A thermite reaction with heat generation can be initiated by a weak ignition source

Molten metal in contact with moisture or certain oxides iron, copper)
Moisture trapped in molten metal can be explosive. Contact with molten
aluminum with certain oxides (lead, iron, copper,, bismuth and some other
metals) an initiate a thermite reaction

Protective equipment and precautions for fire fighters

Fire fighters should wear NIOSH approved positive pressure, self contained breathing apparatus and full protective clothing

Hazardous combustion

Products

No hazardous decomposition products

Fire fighting Equipment

Use gentle surface application of Class D extinguishing agent or **dry** inert granular (sand) to cover the fire and ring the burning material. If impossible to extinguish ring the burning material let fire burn out. Apply extinguishing agent to NOT stir up dust

Explosion data Sensitivity to

Mechanical impact

Not applicable

Sensitivity to Static discharge

Prevent static discharge when dust explosion is possible

6. ACCIDENTAL RELEASE MEASURES

Personal precautions

Avoid generating dust. Avoid inhalation of dust. Avoid inhalation of fume from molten metal. Avoid contact with sharp edges or molten metal. Molten, heated and cold aluminum look alike. Do not ouch unless you KNOW it is cold. Use personal protective equipment as noted elsewhere in this SDS

Environmental precautions

Reuse or recycle material

7. HANDLING AND STORAGE

Evacuation procedure

Persons not wearing personal protective equipment should be excluded from the spill Area until clean up is complete

Spill or leak procedure

Collect scrap for recycling. IF MOLTEN Use DRY sand to contain flow of metal. All tools and containers must be preheated or coated, rust free and approved for use Allow spill to cool before remelting as scrap

Handling

Keep material dry. Avoid generating dust AVOID contact with sharp edges or hot metal. Hot and cold aluminum are not visually different U se personal protective gear as noted in this document.

Storage

Store in DRY place.

Processes That generate Dust or Fines

If processing this product can generate dust or fines, obtain following safety procedures equipment guides from the Aluminum Association Bulletin F-1 and NFPA brochure 654

Use non-sparking tools and natural bristle brushes. Cover and reseal partially empty containers. Provide static bonding as needed to prevent static charge buildup during transfer operations Local ventilation and vacuum systems must be designed to handle explosive dusts

Do not let small chucks, dust or fines to contact water. Avoid ignition sources. Maintain good housekeeping dust accumulation on floors, beams of ledges is an explosion hazard DO NOT USE DRY VACCUMS OR ELECTROSTATIC PRECIRPTATORS DO NOT mingle fines from iron or other metal oxides with aluminum fines

Requirements for Remelting Scrap Molten metal and water and can be an explosive. DRY all ingot or scrap before charging

Material or ingot

Preheat large items before charging filled space into a where molten aluminum traps water. Water and other contaminants can cause explosions. Tools must be preheated or coated. Rust free and approved for use. Any hot or molten that MAY contact surfaces concrete) must be coated during melting operations all material must be inspected prior to charging and surface contaminant removed (ice, snow, grease, or contaminants caused by storage)

Store ingot and sow in heated areas with cracks or cavities pointed down preheat and dry large items before charging into a partially filled furnace. Fine turnings and dust from processing is readily ignitable in furnaces that previously had been used to melt lead, bismuth or other low melting point metals

Dross Handling

Small amounts of beryllium maybe present. This beryllium in dross does not present a health hazard during processing however can be accumulated in dross.

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8. EXPOSURE CONTROLS/PERSONAL PROTECTION

Engineering controls

Dust and fumes must be controlled by an explosion proof ventilation system designed to handles combustible particulates see NFPA bulletin 645

Personal protective Equipment

Eye/face protection

Wear safety glasses with side shields. A face shield must be worn around molten

metal

Thermal hazards

Skin & body protection Around molten metal wear heat resistant glove and Flame retardant clothing Contact with molten material can cause thermal burns. Hot aluminum does not glow red when heated. Wear gloves to protect from burns. Wear flame retardant clothes

Respiratory protection If dust or fumes from processing is present wear respirators. If lead suspected sue special lead respirator

Hygiene measures

Handle following good hygiene and safety practices. Wash hands before eating,

smoking or drinking

General

Personnel handling molten aluminum should wear protective clothing, polycarbonate face shields, fire resistant jackets and pants, spats, snoods, flame resistant hard toed shoes etc. NO synthetic fiber underwear or other clothing

Sample to establish lead levels

Personnel who handle and work with metal use utilize primary protective clothing, fire resistant jackets, poly carbonate face shields, leggings, spats, etc. Synthetic material should never be worn even as under garments.

Sampling to establish lead level exposure is needed where exposure to lead level exposure is possible. See OSHA standard 29 CFR 1910.1025 for specific precautions and requirements.

EXPOSURE CONTROLS/ PERSONAL PROTECTION (Continued) OSHA

OSHA			
Components	Туре	Value	Form
Cobalt (CAS 7440-16-41)	TWA	0,1mg/m3	Dust and fume
Aluminum (CAS 7429-90-5)	TWA	5 mg/m3	Respirable fraction
		15mg/m3	Total dust
Chromium (CAS 7440-47-3)	TWA	1mg/m3	
Copper (CAS 7440-50-8)	TWA	1mg/m3	Dust and mist
		0.1mg/m3	fume
Manganese (CAS7439-96-5)	Ceiling	5 mg/m3	fume
Silicon (CAS 7440-21-3)	TWA	5 mg/m3	Respirable fraction
		15mg/m3	Total dust
Aluminum Oxide Non fibrous	TWA	5 mg/m3	Respirable fraction
(CAS1644-28-1)			
Chromium (II) Compound	TWA	0.5mg/m3	(as Cr)
Chromium (III) Compound	TWA	0.5mg/m3	(as Cr)
Chromium (VI) (CAS 18540-20-9)	TWA	0.0025mg/m3	Action level as Cr VI
Cobalt Compound	TWA	0.01mg/m3	Metal dust/fume
Iron Oxide (CAS 1309-37-1)	TWA	10mg/m3	fume
Lead Compound Inorganic	TWA	0.05mg/m3	As Pb
		0.03mg/m3	Action level Pb
Manganese Compound	Ceiling	5mg/m3	fume
Nickel Compound Insoluble	TWA	1mg/m3	As Ni
Nitric Oxide (CAS 10102-43-9)	TWA	30mg/m3	
Ozone (CAS 10028-15-6)	TWA	25ppm	
		0.2mg/m3	
Lead (CAS 7439-92-1)	TWA	5mg/m3	fume
Lead (CAS 7439-92-1)	gog grapus menerus kasa na kada meneripa meninka sa da a nema semusa na kanana sa pada menerus kada menerus ka	15mg/m3	Total dust
Lead (CAS 7439-92-1)	TWA	0.05mg/m3	
Cadmium (CAS 7440-43-9)	TWA	0.005mg/m3	
Chromium (CAS 7440-47-3)	TWA	0.05mg/m3	

AGHIH Threshold limits

CAS	E 1 24 E 24		
	TWA	VALUE	FORM
1344-28-1	TWA	1 mg/m3	Respirable fraction (as Al)
7440-43-9	TWA	0.01 mg/m3	
		0.002 mg/m3	Respirable fraction
7440-47-3	TWA	0.5 mg/m3	
7440-48-4	TWA	0.02 mg/m3	
7440-50-8	TWA		Dust & Mist
		1 mg/m3	
7440-50-8	TWA	0.2 mg/m3	Fume
7439-92-1	TWA	0.05 mg/m3	
7439-96-5	TWA	0.1 mg//m3	Inhalable fraction
	TWA	0.02 mg/m3	Respirable fraction
	TWA	1.5 mg/m	Inhalable fraction
	7440-43-9 7440-47-3 7440-48-4 7440-50-8 7440-50-8 7439-92-1	7440-43-9 TWA 7440-47-3 TWA 7440-48-4 TWA 7440-50-8 TWA 7440-50-8 TWA 7439-92-1 TWA 7439-96-5 TWA 7440-02-0 TWA	7440-43-9 TWA 0.01 mg/m3 0.002 mg/m3 7440-47-3 TWA 0.5 mg/m3 7440-48-4 TWA 0.02 mg/m3 7440-50-8 TWA 1 mg/m3 7440-50-8 TWA 0.2 mg/m3 7439-92-1 TWA 0.05 mg/m3 7439-96-5 TWA 0.1 mg//m3 7440-02-0 TWA 0.02 mg/m3

9. PHYSICAL & CHEMICAL PROPERTIES

Form

Solid

Color

Silver

Odor

None

Auto-ignition temp

none

Boiling point

2917 °C

Density

2.50 - 3.12 g/cm³

Melting point

899.6° C

pH

Not applicable

Solubility in water Not applicable

10. CHEMICAL STABILITY & REACTIVITY INFORMATION

Chemical stability

Stable under normal conditions of use, storage and transport as shipped

Conditions to avoid

Chips, fines dust and molten metal are much more reactive with: water, generates flammable hydrogen gas and heat. The finer the particles the faster the rate.

Molten metal can react violently with water if the water is entrapped Heat: Oxidizes at a rate dependent on temperature and particle size

Possible hazardous

Reactions

Incompatible materials

Hazardous polymerization does not occur

Chips, fines, dust and molten metal are much more reactive with:

Strong oxidizers: violent reaction with lots of heat. Will explode violently

With nitrates such as ammonium nitrate or fertilizers contain nitrate when heated or

molten

Acids and alkalis: reacts to generate flammable / explosive hydrogen gas

Halogenated compounds can react with fines or molten metal Iron oxide and other metal oxides can create a Thermite reaction

Iron oxide can react with molten aluminum without an external ignition source

Iron powder and water form s an hydrogen gas when heated to 1470°F

Thermite explosions have occurred in furnaces that had previously melted lead, bismuth or other low melting point metals Thermite reactions can occur with oxides

of lead, copper, iron bismuth and some others

Hazardous decomposition

products

no decomposition products known

11. TOXICOLOGICAL INFORMATION

Aluminum dust and fines and fume: Low health risk by inhalation considered biologically inert

Cadmium dust fumes and mist: Can cause severe irritation of the respiratory tract. Acute overexposure can cause metal fume fever, inflammation of lung tissue Can cause lung damage (pulmonary edema) If contains cobalt or nickel may produce allergic reaction Heating above melting point releases metal oxides which may cause metal fume fever by inhalation as well as muscle pain. Long term breathing may cause lung or nasal cancer or chronic lung disease

Chronic exposure to low levels of manganese dust/fume can result in a disease similar the Parkinson's disease, gait impairment muscle spasms and behavioral changes

Lead may damage kidney function, Blood forming system and reproductive system

Cobalt may irritate eyes, skin and/or respiratory tract. Skin contact may cause allergic reactions. Overexposure may cause asthma, pulmonary fibrosis and may harm the heart muscle.

12. ECOLOGICAL INFORMATION FCOTOXICOLOGICAL DATA

Components		Species	Test Results
Aluminum (CAS 7429-90-5)			
Aquatic			
Cladocera	LC50	Water flea	3.5mg/l 24 hours
Fish	LC50	Trout	0.31 mg/l 96 hours
Iron (CAS 7439-89-6)			
Crustecea	LC50	Shrimp	100+ mg/l 48 hours
Fish	LC50	Catfish	>500 mg/; 96 hours
Aluminum oxide (non-fibrous)			

(CAS 1334-28-1)

Aquatic toxicity Not harmful to aquatic organisms

Persistence Not biodegradable

Bioacculation Due to low water solubility not likely

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13. DISPOSAL CONSIDERATIONS

Reuse or recycle material when possible. If reuse or recycling is not **Disposal instructions**

possible disposal must be made according to local or other regulations

Not applicable Waste from residues/

Unused products Not applicable

Mobility in environment Not applicable

14. Transport Information

General Shipping Information

Basic shipping requirements

UN number

not regulated Proper shipping name

Hazard class

Packing group

When "not regulated" enter the proper freight classification, SDS number and Product name onto the shipping papers

Disclaimer

This section provides basic classification information and information re: special modal regulations, environmental hazards and special precautions

15. REGULATORY INFORMATION (United States)

SEE Toxic Substances Control Act (TSCA) Inventory

16. OTHER INFORMATION

requirements of Section This product contains the following toxic chemicals subject to then reporting 313 of the Emergency Planning and Community Right-to-Know Act of 1986 and 40 CFR 372

CAS Number	Chemical Name Aluminum (fume or dust only)	Percent By Weight [a,b]
7429-90-5	·	[a,b]
7429-47-3	Chromium	• • •
7440-50-8	Copper	[a,b]
7439-92-1	Lead	[a,b]
7439-95-5	Manganese	[a,b]
7440-02-0	Nickel	[a,b]
7440-66-6	Zinc ngredient/s(Sections 2 & 3) identity in	[a,b] Iformation by weight perc

See section Hazardous ingredient/s(Sections 2 & 3) identity information by weight percentage

[b] Must be adjusted for the fraction of the material which exists as fume or dust

16. OTHER INFORMATION

References:

Aluminum Association Bulletin F-1 'Guidelines for Handling Aluminum Flnes. Generated during Various Aluminum Fabricating Operations The Aluminum Association, 1525 Wilson Boulevard. Suite 600, Arlington VA 22209. www.aluminum.org

American Conference of Governmental Industrial Hygiests. Threshold Limit Values and Biological Exposure Indices Cincinnati 2015

Bretherick, Handbook of Reactive Chemical Hazards, Butterworth's 1995

Merck & Co. (Royal Surety of Chemistry) The MERCK Index 17th Edition, Rahway, NJ 2015

Lewis, Richard J., Sr. Hazardous Chemicals Desk Reference 3RD edition, Van Nostrand Reinhold, New York 1973 Lewis, Richard J., Sr. Hazardous Chemicals Desk Reference CDROMS edition John Wiley 1998

National Fire Protection Association, National Fire Codes. Manual of Hazardous Chemical Rreactions Quincy MA 2015

Plunkett, E. R. Handbook of Industrial Toxicology: Chemical Publishing Co, New York 1976

Sax, M. Irving: Dangerous Properties of lindustrial Materials CDROM edition, John Wiley New YORK 2112

US Dept of Health & Human services, NIOSH Pocket Guide to Chemical Hazards Cincinnati, 2010

US Dept of Health & Human services, NIOSH Register of Toxic Effects of Chemical Substances 2014

US Dept of Labor, OSHA Regulations 29CFR1910.1000 -1910-1200 January 2015

US Environmental Protection Agency LIST OF lists EPA 550-b3 Washington DC 2015

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